Problem'8. Consider a large tank holding 1000 L of water into which a brine solution of salt begins to flow at a rate of 6 L/min. The solution inside the tank is kept well stirred and is flowing out of the tank at a rate of 6 L/min. If the concentration of salt in the brine entering the tank is 1 kg/L, determine the concentration of salt network.

If m(t) is the mass of salt in the tank at time  $t \Rightarrow C(t)$  is the concentration of salt in the tank of the same time, where  $C(t) = \frac{m(t)}{1000}$ . The rate of change of m is

$$m'(t) = 6 - \frac{6m}{1000}; m(0) = 0$$

this is a linear IVP:  $m' + \frac{3}{500}m = 6$ ,  $\mu(t) = e^{\int \frac{3}{500} dt} = e^{3t/500}$ , then  $(e^{3t/500}m)' = 6e^{3t/500}$ ; Integrating,  $e^{3t/500}m = 1000e^{3t/500} + C \Rightarrow m(t) = 1000 + ce^{-3t/500}$ . Use the initial condition:  $o = m(o) = 1000 + C \Rightarrow C = -1000$  and  $m(t) = 1000(1 - e^{-3t/500})$ , hence  $m(1) = 1000(1 - e^{-3/500})$ .

Problem 9. Suppose that the temperature of the cup of coffee obeys Newton's law of cooling. If the coffee has a temperature of 200 degrees when freshly poured, and one minute later has cooled to 190 degrees in a room at 70 degrees, determine when the coffee reaches a temperature of 150 degrees.

Solution: 
$$T(0|=200 \text{ and } T_{z}=70 \Rightarrow T(t)=70 + (200-70)e^{-kt}=70+130e^{-kt}; 190=T(1)=70+130e^{-k} \Rightarrow 120=130e^{-k} \text{ and } e^{-k} = \frac{12}{13} \Rightarrow T(t)=70+130(\frac{12}{13})^{t}.$$
 Need to solve for t:  $150=70+130(\frac{12}{13})^{t} \Rightarrow (\frac{12}{13})^{t} = \frac{8}{13}; \Rightarrow t \ln \frac{12}{13} = \ln \frac{8}{13} \Rightarrow t = \frac{\ln \frac{8}{13}}{\ln \frac{12}{13}}.$